11. Carbon and its compounds



UNIT - 11 CARBON AND ITS COMPOUNDS

I. Choose the best answer

1.	The molecular formula	of an open chain organ	nic com	pound is C_3H_6 .	The class of t	the compound is
	a) alkane	b) alkene	c) alk	yne	d) alcohol	
2.	The IUPAC name of an	the IUPAC name of an organic compound is 3-Methyl butan-1-ol. What type compound it is?				npound it is?
	a) Aldehyde	b) Carboxylic acid	c) Ket	tone	d) Alcohol	[SEP - 2021]
3.	The secondary suffix used in IUPAC nomenclature of an aldehyde is					
	a) - ol	b) – oic acid	c) - al		d) - one	
4. Which of the following pairs can be the successive members of a homologous series?					series?	
	a) C_3H_8 and C_4H_{10}	b) C ₂ H ₂ and C ₂ H ₄	c) CH	4 and C ₃ H ₆	d) C ₂ H ₅ OH	and C ₄ H ₈ OH
5.	$C_2H_5OH + 3O_2 \rightarrow 2CO$	$O_2 + 3H_2O$ is a				[SEP-2020]
	a) Reduction of eth	anol	b) Cor	nbustion of eth	anol	
	c) Oxidation of eth	anoic acid	d) Ox	idation of etha	nol	
5.	Rectified spirit is an ac	queous solution which	contair	ıs about	of ethanol	. [MAY - 2022]
	a) 95.5 %	b) 75.5 %	c) 55.	5 %	d) 45.5 %	
7.	Which of the following	g are used as anaesthe	etics?			
	a) Carboxylic acids	s b) Ethers	c) Est	ers	d) Aldehyde	es
3.	TFM in soaps represen	ntsconte	ent in so	oap.		
	a) mineral	b) vitamin	c) fatt	y acid	d) carbohyd	rate
9.	Which of the following statement is wrong about detergents?					
	a) It is a sodium sal	t of long chain fatty ac	ids	b) It is sodium	m salts of sulp	phonic acids
	c) The ionic part in	a detergent is –SO ₃ I	Na ⁺	d) It is effect	ive even in ha	ard water

II. Fill in the blanks

- 1. An atom or a group of atoms, which is responsible for chemical characteristics of an organic compound is called **functional group**.
- 2. The general molecular formula of alkynes is $\underline{C_n H_{2n-2}}$.
- 3. In IUPAC name, the carbon skeleton of a compound is represented by **root word**(root word/prefix/suffix)
- 4. (Saturated / Unsaturated) <u>Unsaturated</u> compounds decolourize bromine water.
- 5. Dehydration of ethanol by conc. sulphuric acid forms ethene. (ethene/ ethane)
- 6. 100 % pure ethanol is called **absoulte alcohol**.
- 7. Ethanoic acid turns blue litmus to red.
- 8. The alkaline hydrolysis of fatty acids is termed as **saponification**.
- $9. \ \ Biodegradable\ detergents\ are\ made\ of\ \underline{\textbf{straight}}\ (branched\ /\ straight)\ chain\ hydrocarbons.$

III. Match the following

[PTA - 2]

Column I	Column II	Answer
1. Functional group –OH	Benzene	1. Alcohol
2. Heterocyclic	Potassium stearate	2. Furan
3. Unsaturated	Alcohol	3. Ethene
4. Soap	Furan	4. Potassium stearate
5. Carbocyclic	Ethene	5. Benzene

IV. Assertion and Reason

Answer the following questions using the data given below.

- i) A and R are correct, R explains the A
- ii) A is correct, R is wrong.

iii) A is wrong, R is correct.

- iv) A and R are correct, R doesn't explains A.
- 1. Assertion (A): Detergents are more effective cleansing agents than soaps in hard water. [PTA 4]

Reason (R): Calcium and magnesium salts of detergents are water soluble.

Ans. (i)

A and R are correct, R explains the A.

- **2. Assertion** (**A**) : Alkanes are saturated hydrocarbons.
 - **Reason** (**R**) : Hydrocarbons consist of covalent bonds.

Ans. (iv)

A and R are correct, R doesn't explains A.

V. Short answer questions

1. Name the simplest ketone and give its structural (or) molecular formula.

[PTA -2]

Simplest ketone

: Acetone

Structural formula : CH₃-CO-CH₃

IUPAC Name : Propanone

$$\begin{bmatrix} H & O & H \\ -C & -C & -C - H \\ 1 & 1 & 1 \\ H & H \end{bmatrix}$$

2. Classify the following compounds based on the pattern of carbon chain and give their structural formula: (i) Propane (ii) Benzene (iii) Cyclobutane (iv) Furan [PTA – 1, MAY-2022]

Name	Classification	Molecular formula	Structural formula
i) Propane	Acyclic (or) open chain compounds	C ₃ H ₈	H H H H - C - C - C - H H H H
ii) Benzene	Aromatic compound	$\mathrm{C}_6\mathrm{H}_6$	HC CH OT
iii) Cyclobutane	Alicyclic compound	C ₄ H ₈	H ₂ C—CH ₂ or
iv) Furan	Heterocyclic compound	C ₄ H ₄ O	HC_CH or O

3. How is ethanoic acid prepared from ethanol? Give the chemical equation.

Ethanoic acid is prepared from oxidation of ethanol in the presence of alkaline potassium permanganate or acidified potassium dichromate.

> KMnO₄/OH CH₃CH₂OH CH₃COOH + H₂O 2[0] Ethanol Ethanoic acid

4. How do detergents cause water pollution? Suggest remedial measures to prevent this pollution.

Detergents having branched hydrocarbon chain are not fully biodegradable by microorganisms in water. So, they cause water pollution. [PTA-3]

<u>Remedy:</u> We can use straight hydrocarbon chains, which can be easily degraded by bacteria.

5. Differentiate soaps and detergents.

[SEP - 2020, PTA - 3, MDL - 19]

Soaps	Detergents		
1. Sodium salt of long chain fatty acid	1. Sodium salts of sulphonic acids.		
2. Ionic part is –COO [–] Na ⁺	2. Ionic part is – SO ₃ ⁻ Na ⁺		
3. It is less effective in hard water.	3. 4. It is effective even in hard water.		
4. It forms a scum in hard water.	5. It does not form a scum in hard water.		
5. Poor foaming capacity.	6. Rich foaming capacity.		
6. Biodegradable.	7. Mostly non-biodegradable.		

VI. Long answer questions

1. What is called homologous series? Give any three of its characteristics. (write any 3)

Organic compounds having same general formula and similar chemical properties in which the successive members differ by a -CH₂ group is called homologous series.

Ex: Methane CH₄ Ethane CH₃-CH₃ Propane CH₃-CH₂-CH₃

Characteristics of homologous series:

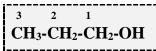
- i) Each member differs from its preceding or succeeding by methylene (-CH₂) group.
- ii) All members contain same elements and functional group.
- iii) They are represented by a general molecular formula. Ex: Alkanes C_nH_{2n+2} .
- iv) Members in each series show regular gradation in their physical properties.
- v) Chemical properties are similar.
- vi) All the members can be prepared by a common method.

2. Arrive at, systematically, the IUPAC name of the compound: CH₃-CH₂-CH₂-OH. [TB:162]

Step1: Longest chain has 3 carbon atoms. ∴ Root word is '**prop**'.

Step2: It has single bond. : Primary suffix is "ane".

Step3: Functional group is alcohol (-OH).



Step 4: Locant number of –OH is 1 ∴ Secondary suffix is '1-ol'

∴ The name of the compound is **Propan** $-\sqrt{1 - ol}$

3. How is ethanol manufactured from sugarcane?

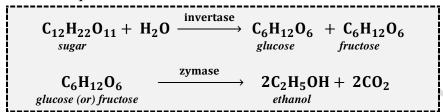
❖ Ethanol is manufactured by fermentation of molasses.

Steps in conversion of molasses to ethanol:

- (i) Molasses is diluted with water to bring the concentration of sugar to 8 to 10%.
- (ii) Addition of Nitrogen Source: It is fortified by adding ammonium sulphate/phosphate.

(iii) Addition of yeast:

- Solution is collected in large 'fermentation tanks'
- ❖ Yeast is added and kept at 303 K for few days.
- ❖ During this period, invertase and zymase in yeast, converts sucrose into ethanol.
- Fermented liquid is called as 'wash'.



(iv) Distillation of Wash:

'Wash' contains 15 to 18% alcohol. It is subjected to fractional distillation.

Rectified spirit: It contains 95.5% ethanol & 4.5% of water. It is the main fraction of 'wash'. **Absolute alcohol:** Rectified spirit is refluxed over quicklime for 5 to 6 hours and then allowed to stand for 12 hours. **Pure/absolute alcohol** (100%) is obtained.

4. Give the balanced chemical equation of the following reactions:

(i) Neutralization of NaOH with ethanoic acid.

[PTA-6]

(ii) Evolution of carbon dioxide by the action of ethanoic acid with NaHCO₃.

- (iii) Oxidation of ethanol by acidified potassium dichromate. [PTA-6] (or) Write a reaction which is used for the identification of alcohol., [SEP-2020]
 - ❖ Ethanol is oxidized to ethanoic acid in presence of acidified potassium dichromate.
 - \diamond Orange color of $K_2Cr_2O_7$ is changed to green colour. Thus, it is used to identify alcohol.

$$\begin{array}{c} CH_3CH_2OH \xrightarrow{K_2Cr_2O_7/H^+} CH_3COOH & + H_2O \\ & & ethanol & & ethanoic acid & & & \end{array}$$

(iv) Combustion of ethanol.

$$\begin{array}{ccc} CH_3CH_2OH + 3O_2 & \longrightarrow & 2CO_2 & + & 3H_2O \\ & & & & \\ ethanol & & oxygen & carbon dioxide & water \end{array}$$

- 5. Explain the mechanism of cleansing action of soap. [PTA-6] (or) Explain how micelles formation take place with a diagram when soap is added to water? [PTA-5]Structure of Soap:
 - ***** *Polar end:* It is hydrophilic (water loving). Short head with carboxylate group (-COONa)
 - * Non-polar end: It is hydrophobic (water hating). Long tail of hydrocarbon chain.

Cleansing action of Soap:



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- ❖ When a soap is dissolved in water, molecules join together as clusters called micelles.
- ❖ Dirt is surrounded by non-polar end.
- ❖ Polar end makes micelles soluble in water.
- * Thus, dirt is washed away with the soap.

VII. HOT questions

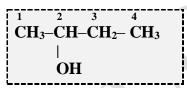
- 1. The molecular formula of an alcohol is $C_4H_{10}O$. The locant number of its -OH group is 2.
 - (i) Draw its structural formula.

[PTA-1]

(ii) Give its IUPAC name.

[PTA – 1]

- Step 1: Chain has 4 carbon atoms. : Root word is 'But'
- Step 2: It has single bond. : Primary suffix is 'ane'.
- *Step 3:* Functional group is alcohol (–OH).



Step 4: Locant number of -OH group is 2. : Secondary suffix is '2-ol'

∴ The name of compound is **Butan-2-ol**

(iii) Is it saturated or unsaturated?

[PTA-1]

Butan-2-ol is saturated as it has only single bonds

- 2. An organic compound 'A' is widely used as a preservative and has the molecular formula $C_2H_4O_2$. This compound reacts with ethanol to form a sweet smelling compound 'B'. [PTA 5]
 - (i) Identify the compound 'A' and 'B'.

Compound (A) is Ethanoic acid or Acetic acid. Its structural formula is CH₃COOH.

(ii) Write the chemical equation for its reaction with ethanol to form compound 'B'.

$$\begin{array}{c|c} CH_3COOH & + & CH_3CH_2OH \xrightarrow{Con.H_2So_4} CH_3COOCH_2CH_3 + H_2O \\ \hline \\ \textit{ethanoic acid}(A) & \textit{ethanol} & \textit{ethanol} & \textit{ethyl ethanoate } (B) \end{array}$$

(iii) Name the process (or) chemical reaction.

This process is esterification.

A → Ethanoic acid
B → Ethyl ethanoate
Process → Esterification